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Development of the Periodic Table . On the line at the left, write the letter of the contribution that each chemist made to the periodic table. Each letter will be used only once. 1. Dmitri Mendeleev a. arranged elements by atomic number b. KM Page ...

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3—3 Review and Reinforcement Modern Atomic Theory Complete the following table. 1. 2. 3. If the statement is true, write "true." If it is false, change the underlined word or words to make the

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statement true. Write your answer on the line provided. 4. Moseley discovered that all atoms of an element have the same number of neutrons in ...

Name Date Class 3—3 Review and Reinforcement

2 - 3 Review and Reinforcement-Matter-
Type your answers in Red Complete the following sentences by filling in the appropriate word from the list below.
Gas Liquid Solid Plasma Matter Energy
Physical Chemical Evaporation Matter is anything that has mass and volume. 1. The two states of matter that occupy a definite volume are solid and liquid. 2. gas and plasma are the other two states of matter.

4. 2\0133 Review and Reinforcement - Matter - 4 23 ...

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Check the following equalities for errors. If an answer is correct, write "correct" in the space provided. If an answer is incorrect, rewrite it to make it correct.

21. $= 4.563 \times 10^7$ 22. $0.000\ 253 = 2.53$

$\times 10^{-3}$ 23. $= 68.05 \times 10^8$ Solve each of

the following problems as directed. Show all your work. 24. 25. 26. 34 An unknown liquid has a mass of ...

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2 Answer.Key 2-1 Explore, page 5 The original source of energy for this activity would be the stored chemical energy in

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Answer

the ... 2-1 Review and Reinforcement, pages 8-9 . 1. energy 2. radiant 3. chemical potential 4. joule 5. created, destroyed 6. natural gas 7. kinetic . 8. f . 9. c 10. b 11. a . 12. e .

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3—3 Review and Reinforcement (continued) Use the periodic table to determine how many protons, neutrons, and electrons are present in each of the following atoms. Write your answers in the spaces provided. Atom 12. iodine-125 13. potassium-39 14. iron-56
Protons Neutrons Electrons Write the chemical symbol for each of the ions

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