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Heat Capacities The

generic heat capacity of

$C_p = a + b/T + c/T^2$

The heat capacities of various

elements and compounds are

$C_p = a + b/T + c/T^2$

$a = 8.54, b = 21.30, c = 1.51 \dots$

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much wider applicability. In material science, one is normally not that interested in heat and work, but interested more the state of matter and how things might change when mixed, heated, pressurized, etc. Some important effects are chemical reactions (such as oxidation), formation of solutions, phase transformations.

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