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Chapter 20:

Radioactivity and
Nuclear Chemistry.

radioactivity.

phosphorescence.

alpha (α) decay. beta
(β) decay. the emission
of subatomic particles
or high-energy

electromagn.... The
long-lived emission of
light that sometimes
follows the ab.... the
form of radioactive
decay that occurs
when an unstable
nuc....

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20: Oxidation-

Reduction Reactions 49

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Chapter 17-18 Review
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Guide Rate of Reaction

Lab Chapter 20 SG

20.1 Determining

Oxidation Numbers SG

20.2 & 20.3 What is

Oxidizing and

Reducing? Balancing

with Oxidation

Numbers Using Half

Reactions Chapter 20

Supplemental Problems

Chapter 20 Assessment

Chapter 20 Worksheet:

Redox Chapter 25 SG

25.1-25.2 SG 25.3-25.4

Balancing ...

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difference between acids and bases. An acid is a substance that donates an H^+ ion to another chemical species called a base. A base is a substance that accepts (combines with) an H^+ ion.. If you need to know the concentration of the H^+ ion in solution, you can do so by representing the H^+ molarity, $[H^+]$.

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Chapter 2 Students
should be able to...

Describe Dalton's atomic theory. Count protons, neutrons, and electrons in an isotope. Know properties of protons, neutrons, and electrons including: symbol, mass number, location, and relative charge. Describe the Rutherford experiment, and how it proved the existence of the nucleus.

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