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first A. Democritus. image of individual atoms. B. John Dalton
____ 2. He confirmed the existence of the . neutron. C. J.J. Thomson
____ 3. He developed the beginnings of the modern D. James
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CHAPTER 5, Atomic Structure and the Periodic Table(continued)
An atom of neon-22 has two more neutrons in its nucleus than an atom of neon-20. isotopes true b c a carbon-12; 12 amu false
The values of atomic masses measured in grams are inconveniently small and impractical to work with. Guided Reading and Study Workbook43

5 ATOMIC STRUCTURE AND THE PERIODIC TABLE

CHAPTER 5, Atomic Structure and the Periodic Table(continued)
Just How Small Is an Atom? (page 108) 6. Suppose you could grind a sample of the element copper into smaller and smaller

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particles. The smallest particle that could no longer be divided, yet still has the properties of copper, is ____ . 7.

5 ATOMIC STRUCTURE AND THE PERIODIC TABLE

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CHEMISTRY NOTES Chapter 5 Atomic Structure and the ...

Chapter 5 Atomic Structure. Franklin. Crookes. LaVoisier. John Dalton. discovered electricity in lightning storm. developed cathode ray tube. developed the law of conservation of mass, particles not creat.... combined Democritus' writings, the work of

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LaVoisier, crookes,....

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ATOMIC STRUCTURE AND THE PERIODIC TABLE CHAPTER 4
WORKSHEET PART A Given the following isotopes, determine the atomic number, the mass number, the number of protons, electrons and neutrons. Isotope Symbol ^{131}I ^{238}U ^4He ^{81}Br ^{36}Kr ^{37}K Atomic Number Mass Number Protons Electrons Neutrons Isotope Name PART B

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