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Aldehydes can be
oxidized to carboxylic
acids; ketones cannot.

The Tollens test and
Benedicts test take
advantage of this. Both
tests involve reagents
which react with
aldehydes to produce a
visible product. No
reaction occurs with
ketones. 18 See the
solution in the text

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sour and cause certain
dyes to change color.
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feel soapy. • Arrhenius
concept of acids and
bases: • An acid is a
substance that, when
dissolved in water,
increases the

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concentration of H^+
ions.

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Name: Answer Key Due
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pts) Circle all the
aromatic molecules.
Put an "X" through the
antiaromatic
molecules. Leave alone
all nonaromatic

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molecules. (1 or 2
incorrect, ok, but after
that start taking partial
points off)

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transfers of energy as

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heat that accompany
ch... The energy
absorbed or released
as heat in a chemical
or physi.... A measure
of the average kinetic
energy of the particles
in a....

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...

The probability for all
the particles to be on
one side is $\frac{1}{32}$. $\frac{1}{32}$.
This probability is
noticeably lower than
the $\frac{1}{8}$ $\frac{1}{8}$ result for

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the four-particle system. The conclusion we can make is that the probability for all the particles to stay in only one part of the system will decrease rapidly as the number of particles increases, and, for instance, the probability for all molecules of gas to ...

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to the Problems IUPAC
names for each

compound, Specify the

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configuration in (c). O

O C₆H₅\CHO(a)

7<\CHO (b) (C) 5CH₃

(R)-2-Phenylpropanal1

fi-CyclohexanedionePro

blem 16.1 Write the3,3

-DimethylbutanalProble

m 16.2 Write structural

formulas for all

aldehydes with the

molecular formula

C₆H₁₂O and give each

its IUPAC name.

Brown&Foot

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electrons are shared equally by the bonded atoms, resulting in a balanced distribution of electrical charge ...

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If the pH of tomato
juice is 4.0, what is the
hydroxide ion

concentration in this
solution? 2. If the pH of

an acid rain storm is
approximately 3.0, how

many times greater is
the (H⁺) in the rain than

in a cup of coffee

having a pH of 5.0? 3.

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