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## **Chapter 15 Water Aqueous Systems**

Chapter 15-Water and Aqueous Systems. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. Joseph\_Wang6. Terms in this set (233) Water is an amazing liquid that

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makes. plant and animal life on Earth Possible. In addition to water forming the Earth's oceans water, it also forms the.

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Chapter 15 Water and Aqueous Systems.

Chapter 15 "Water and Aqueous Systems". The Water Molecule: a

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Review. • Water is a simple tri-atomic molecule, H<sub>2</sub>O. • Each O-H bond is highly polar, because of the high electronegativity of the oxygen (N, O, F, and Cl have high values) • bond angle of water = 105°.

## **Chapter 15 Water and Aqueous Systems**

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Chemistry, Chapter 15, Water and Aqueous Systems. surface tension.

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surfactant. aqueous solution. solvent.  
the inward force or pull that tends to  
minimize the surface ar.... any  
substance that interferes with hydrogen  
bonding between wa.... is water that  
contains dissolved substances.

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10 including work step by step written by community members like you.

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## **Chapter 15 - Water and Aqueous Systems - 15.2 Homogeneous ...**

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Chapter 15 Water and Aqueous Systems 159. SECTION 15.1 WATER AND ITS PROPERTIES (pages 445–449) This section describes the properties of water in the liquid and solid states and explains how hydrogen bonding affects the surface tension and vapor pressure of water. Water in the Liquid State (pages 445–447)

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## **SECTION 15.1 WATER AND ITS PROPERTIES (pages 445-449)**

Chapter 15 "Water and Aqueous Systems". use these activities to yourself study the vocabulary and major concepts presented in this chapter. heterogeneous mixture containing particles that are small enough to

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remain dispersed in the solvent and do not separate on standing.

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Section 15.1 Water and Its Properties  
Polarity is the unequal sharing of [valence]electrons in a covalent bond. Water is a highly polar molecule. Polarity is represented by the Greek symbol " $\delta$



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(Delta)” The polarity of molecules depends on the electronegativity of the elements.

## **Chapter 15 Water and Aqueous Systems by Hanan Akel on ...**

Chapter 15 - Water and Aqueous Systems - 15.1 Water and Its Properties - 15.1 Lesson Check - Page 493: 2 Answer

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The structure of ice is a regular open framework of water molecules arranged like a honeycomb.

## **Chapter 15 - Water and Aqueous Systems - 15.1 Water and ...**

Chapter 15 - Water and Aqueous Systems - 15.1 Water and Its Properties - 15.1 Lesson Check - Page 493: 6 Answer

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The expansion causes pressure to build up that could eventually burst the pipe.

## **Chapter 15 - Water and Aqueous Systems - 15.1 Water and ...**

CaitlinEHogan. Chemistry: Chapter 15: Water and Aqueous Systems Vocabulary. surface tension. surfactant. aqueous solution. solvent. an inward

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force that tends to minimize the surface area of a l.... any substance that interferes with the hydrogen bonding between.... water that contains dissolved substance.

**chapter 15 vocabulary chemistry  
aqueous systems Flashcards ...**

Chapter 15 Water & Aqueous Systems

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Section 15.1 Water & Its Properties  
Water In The Liquid State  
Water is a triatomic molecule  
Oxygen atom covalently bonded to hydrogen(s)  
Polar molecule due to its electronegativity (dipole)  
Bonding angle is approximately 105 degrees, which gives it a bent

**[PDF] Chapter 15 Water And**

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## **Aqueous Systems Id A**

Chapter 15 - Water and Aqueous Systems - 15.2 Homogeneous Aqueous Systems - 15.2 Lesson Check - Page 501:  
12 Answer The forces holding the water molecules in hydrates are not very strong, so the water is easily lost and regained

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**Chapter 15 - Water and Aqueous  
Systems - 15.2 Homogeneous ...**

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Township High School**

Chapter 15 - Water and Aqueous

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Systems - 15 Assessment - Page 510: 28  
Answer Surfactants are wetting agents  
and include common household  
substances such as soap and detergent.

## **Chapter 15 - Water and Aqueous Systems - 15 Assessment ...**

Title: Chapter 15 Review Water and  
Aqueous Systems 1 Chapter 15 Review



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## Answers

Water and Aqueous Systems. Pre-AP Chemistry ; Charles Page High School ; Stephen L. Cotton; 2 Chapter 15 Review. Surface tension is the \_\_\_\_\_. How does the surface tension of water compare with the surface tensions of most other liquids? Which type of mixture(s) exhibit the ...

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Answers

## **PPT - Chapter 15 Review Water and Aqueous Systems ...**

Chapter 15 "Water and Aqueous Systems" Pre-AP Chemistry Charles Page High School Stephen L. Cotton 2. Section 15.1 Water and it's Properties□

OBJECTIVES: -Explain the high surface tension and low vapor pressure of water in terms of the structure of the water

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Answers  
molecule and hydrogen bonding. 3.

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